

CHARACTERISATION AND OSL DATING OF AN ANCIENT BRICK WELL AT PATTARAIPERUMBUDUR, TAMIL NADU, INDIA

KARAKTERIZACIJA IN DATIRANJE STARODAVNEGA OPEČNATEGA VODNJAKA V PATTARAIPERUMBUDURU, TAMIL NADU, INDIJA

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This paper is about characterizing a brick sample obtained from the brick well at Pattaraiperumbudur, a recently excavated site in Tiruvallur, Tamil Nadu. The petro-archaeometry studies of this site included the findings about its elemental constitution, mineralogical composition, thermal changes and physical properties. The elemental constitution was determined using an XRF test, while the mineral composition was found with the help of XRD and FT-IR, showing the chemical bonding at various peaks. A material can undergo different thermal reactions at various temperatures and the effect of temperature on the sample was screened through a TG-DTA analysis. Optical microscopy (OM) is one of the important tests used in petro-archaeological studies, confirming the XRD results. The morphology of the sample was studied with the aid of FE-SEM, which provided an idea about the porous structure and elemental composition. Physical tests of water absorption, porosity and bulk density provided information on the durability of the brick sample. The combined results of TGA and XRD suggest that the firing of the bricks had taken place below 950 °C in an oxidising atmosphere. The chemical composition reveals that the clay contains a low amount of CaO and the flux concentration is greater than 9 %, indicating low refractory. The extracted brick samples were subjected to carbon dating using optically stimulated luminescence (OSL) that found them to be 2100 years old. 3D rendering of the well was created with the help of the H-BIM tool.

Keywords: mineral characterisation, microstructural analysis, OSL dating

V članku avtorji opisujejo karakterizacijo vzorca opeke, vzete iz opečnatega vodnjaka v Pattaraiperumbuduru, kjer so nedavno izvedli geološka izkopavanja v Tiruvallurju, Tamil Nadu, v Indiji. Petro-arheološke študije tega nahajališča obsegajo raziskovanje osnovne kemijske zgradbe, mineraloške sestave, termalnih sprememb in fizikalnih lastnosti izkopanih vzorcev. Kemijsko sestavo vzorcev so določili z rentgensko fluorescenco (XRF), medtem ko so mineraloško sestavo določili z rentgensko difrakcijo (XRD) in Fourierjevo transformacijsko infrardečo spektroskopijo (FT-IR). Material med segrevanjem in ohlajanjem lahko doživi vrsto sprememb. Te so opazovali s pomočjo termo-gravimetrije (TG) in diferencialne termične analize (DTA). S pomočjo optične mikroskopije (OM), ki je ena od osnovnih metod petro-arheoloških raziskav, so potrdili rezultate dobljene z XRD. Morfologijo vzorcev so študirali z vrstično elektronsko mikroskopijo na poljsko emisijo (FE-SEM), ki lahko pomaga pri opisu porozne strukture in kemijske sestave. Fizikalne preiskave, kot so na primer določitev absorpcije vode, poroznosti in volumnske gostote, podajajo informacijo o dobi trajanja oz. starosti vzorcev opeke. S kombiniranjem rezultatov TGA in XRD so avtorji ugotovili, da je bila opeka žgana pod 950 °C v oksidacijski atmosferi. Kemijska analiza je pokazala, da glina vsebuje majhno vsebnost CaO in da je vsebnost talil večja kot 9 %, kar kaže na to, da ima analizirana opeka slabo ognjevzdržnost. Izkopani vzorci opeke so bili datirani z ogljikom oz. uporabo optično simulirane luminescence (OSL). Datiranje je pokazalo, da so vzorci stari 2100 let. 3-D prikaz vodnjaka so obnovili in predstavili s pomočjo programskega orodja H-BIM (angl.; Heritage Building Information Modeling).

Ključne besede: mineraloška karakterizacija, mikrostrukturalna analiza, datiranje z optično simulirano luminescenco (OSL).

1 INTRODUCTION

This paper highlights important correlations between civil-engineering, archaeological and architectural values. The civil-engineering aspect involves the study of the materials and production technology adopted during a construction; the archaeological aspect involves the investigation of the period, in which the structure was built; the architectural aspect tells us about the creative ideas of ancient people. When we look into our past, we

find many vital structures and monuments, which had been constructed in the early centuries. The technology used then is still a mystery to the present generation. So, it is of utmost importance to determine the methods and techniques used in our country for the construction of these ancient heritage structures. This study includes a characterisation of the building materials. The characterisation process includes mineralogical, elemental and textural detailing of the sample.¹ This kind of study is termed as petro-archaeometry.

Bricks are one of the most ancient building materials known to us. The aesthetic aspect of brick work has

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gained a lot of consideration over the years. The historic relevance of these ancient structures or monuments is of utmost importance not only for the analyses and preservation process but also for understanding the technologies used by the ancient people; besides, these structures connect the ancient and modern era. The knowledge of the raw materials used in construction process is very important for the restoration of ancient structures.²

The study of ancient structures has always been crucial in our research area.⁴ The wide research in this area helps us understand the evolution of construction technologies and the living style of our ancestors.⁴ The potentials of the study of heritage science are enhanced with the help of petro-archaeometric studies, analysing the chemical compositions of the materials used for the construction of these structures.^{3,4}

Pattaraiperumbudur is an archaeologically important site excavated in the village of Tiruvallur, Tamil Nadu. Numerous archaeological artefacts from the prehistoric period to the early historic period were found during the excavation. These include pottery, stone tools and bricks, which represent the Palaeolithic culture of the site. Apart from the stone-age tools, the remnants of the iron-age culture were also identified. During the excavation, a brick well was found at a depth of 2.85 m. Based on a visual inspection, a sample brick seemed less porous and comparatively larger than the present bricks, rectangular and orange-red in appearance. No salt efflorescence that reduces a major stress caused to bricks was observed.²

In order to find the technologies used in the manufacturing of building materials, the characterisation of the brick sample was important.⁵ The characterisation included the study of the physical properties, mineral composition, thermal reactions during the baking process and

also the microstructure of the material.^{2,3} The physical properties studied were the water absorption, porosity and bulk density.¹⁰ The water absorption is proportional with the porosity. The chemical composition of the sample was determined with XRF. With the help of the XRF data, the type of clay was determined with respect to whether it was a calcareous or non-calcareous substance,⁶ while a TG-DTA analysis helped us understand the firing temperatures and weight loss due to the thermal activity.^{2,6} Mineralogical analyses were performed with the help of X-ray diffraction.^{2,3,4} Petrographic and microstructural studies were carried out with the help of optical microscopy (OM). The texture level of the vitrification and the pore size were evaluated with the help of FE-SEM.^{1,8} EDS gave the elemental composition for a particular region of the sample.^{3,14} FT-IR helped us study chemical reactions such as decomposition, dihydroxylation and deconvolutions associated with heating.^{9,10,13} OSL dating gave the information about the age of the sample.²¹ Dating can be done using two methods, thermoluminescence and optical luminescence. In this project, optically stimulated luminescence was carried out. Some minerals are capable of storing a small amount of energy derived from the radioactive elements such as uranium, thorium and potassium when they are exposed to radioactive decay. Later, the ejected electrons and holes recombine with the opposite charges when the light is stimulated and thus luminescence of the sample is formed.²³ We also used the terms equivalent dose, which is the amount of energy absorbed by a buried sample, and dose rate, which is the measurement of the energy absorbed every year from the radiation present in

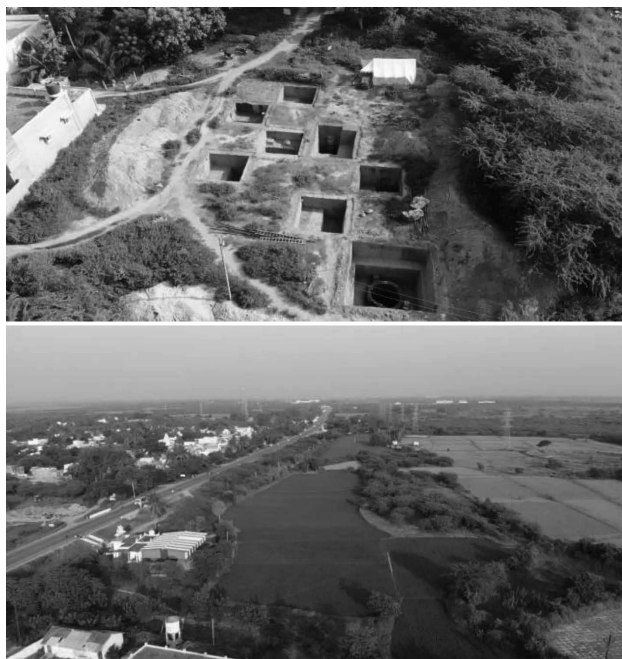


Figure 1: Ariel view of the Pattaraiperumbudur excavated site

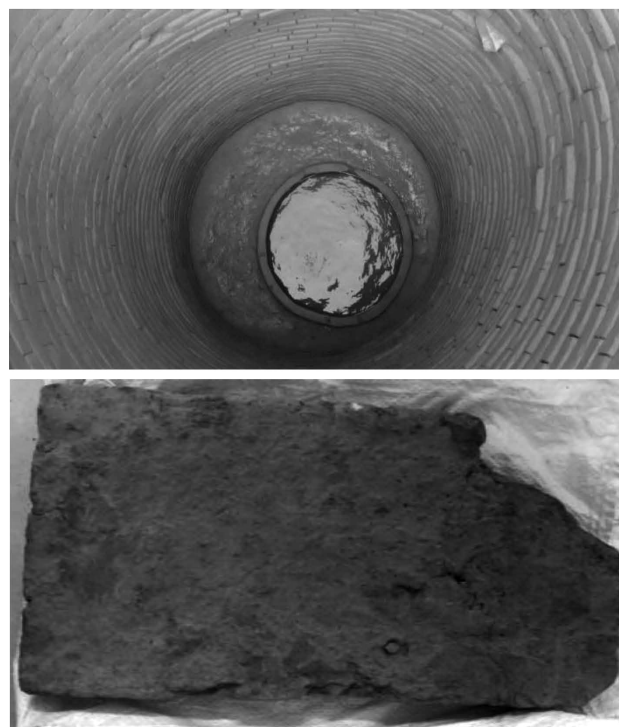


Figure 2: Brick well and the brick sample

the surrounding environment. Dating techniques were mainly used by the department of archaeology for determining the era of the construction of the remains obtained from the excavation site. **Figure 1** shows the ariel view of the site.

2 MATERIALS AND METHOD

2.1 Collection of samples

The site of Pattaraiperumbudur contributes significantly towards the knowledge about ancient settlements. The site lies on the eastern bank of the Kosathalaiyar River. A number of sites on the bank of the Kosathalaiyar River have a great significance for the Indian Palaeolithic archaeology. An archaeological team carried out an excavation between May and September 2018. The sampling of well was done under the guidance of the officials from the Department of Archaeology, Tamil Nadu. A brick sample was removed from the top course. Many trenches with sizes of 10 m × 10 m and 4.25 m × 4.25 m were dug during the excavation at the Pattaraiperumbudur site. The brick well was found in trench B1 at a depth of 2.85 m. From the preliminary examination of the site, information regarding the size, depth and nature of the soil was obtained. The depth from the first course to last course of the well is about 3.91 m. The exterior and interior diameters of the well are 2.60 m and 2.20 m. The number of brick courses is 52 and each course consists of 32 bricks. The average size of a brick is 24 cm × 24 cm × 7 cm. **Figure 2** shows the well and the brick.

2.2 Optical microscopy

The mineral composition was observed using a Zeiss optical microscope. Thin polished sections required for optical microscopy were prepared at Chennai, Tamil Nadu. Petrographic sections were prepared by cutting a slice of the sample with a thickness of 0.03 mm, which was then attached to a glass slide with epoxy resin and covered with a glass slip. This slide is examined under a polarising microscope.

2.3 X-ray diffraction (XRD)

A Siemens D500 Advance diffractometer using Cu-K radiation, equipped with a NaI (TI) scintillation detector was used for the XRD analysis. Samples of about 5–10 mg were finely ground, sieved through a 45-micron sieve and used for the analysis. XRD provided the mineral composition of the brick sample. The Xpert high-score software was used to identify the mineralogical composition.

2.4 Fourier transform infrared spectroscopy (FT-IR)

An IR Affinity-1 using software IR Solution 1.60 with a resolution of 0.5–16 cm⁻¹ was used for the FT-IR

spectroscopy. The instrument has a precision of ±5 cm⁻¹. This investigation was done using the KBr pellet technique, whose wave number ranged from 4000 cm⁻¹ to 400 cm⁻¹ and also by mixing the powdered samples with KBr in the 1:20 weight proportion. A sample of about 5–10 mg was required for the analysis; infrared radiation was applied to it, some of which was absorbed by the sample while some passed through it. Each molecular element has a different infrared pattern; thus, functional organic and inorganic groups can be identified.

2.5 Field-emission scanning electron microscopy (FE-SEM)

A morphological study of the samples was carried out using a Thermo Fisher FEI QUANTA 250 FEG equipped with a Schottky field-emission electron gun as the electron source. It provides a high resolution of 1.2 nm at 30 kV and high vacuum in an operating voltage range of 5–30 kV. The sample used for the analysis was sputter coated with a very thin gold layer, resisting the charging and the sample was magnified in a range of 5–10,000 kX with a spot energy-dispersive spectrometer (EDS) to evaluate the elemental composition of the sample.

2.6 X-ray fluorescence (XRF)

The elemental bulk compositions of the finely ground <45 µm powder as lime-paste reproductions in w/% oxides were obtained using a JEOL JS83201Z X-ray fluorescence spectrometer. About 10 g of powder was inserted into the sample tube of a 3628 Bench-Press Spex Cretiprep disc press machine to form a thin sample plate at a pressure of 20 tons in 2 min. The resulting sample disc was placed into the sample holder of the X-ray fluorescence spectrometer and the analysis of each oxide was performed within an error of 0.01 w/%.

2.7 Thermogravimetric analysis (TGA)

TG-DTA analyses were carried out using a SDT Q600. It has a heating rate of 5–20 °C/minute. Samples can be heated in nitrogen up to 1200 °C. Dried powdered samples of 5–10 mg were used for the analysis, during which the rate of change in the weight of a sample as a function of temperature or time was measured in a controlled atmosphere.

2.8 Optically stimulated luminescence (OSL)

The lab was equipped with a TL-DA-15 luminescence reader and alpha counter to determine the uranium and thorium contents in a sample. The sample preparation was done under subdued-red-light conditions. Blue or green light was used when dealing with quartz and infrared light was used for feldspar. The single-aliquot protocol (SAR) was followed. In the preliminary treatment, first the sun-exposed layers were filed off and crushed in

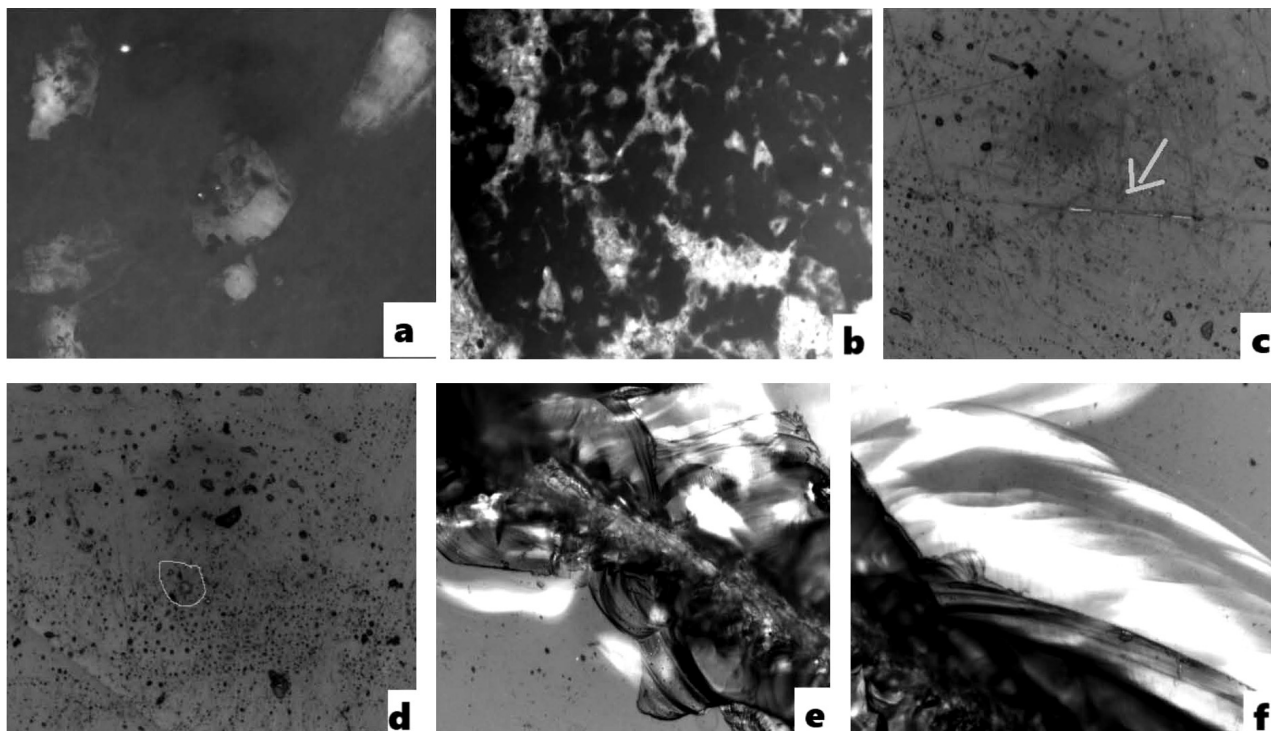


Figure 3: Optical microscopy: a, b) quartz; c) feldspar; e, f) boromuscovite

a beaker where the removal of carbonates and organics was done with chemical solutions. Density separation was done to separate quartz and feldspar; etching was also done to remove the alpha skin. The preliminary treated sample was then fixed to the stainless disc using silicon spray. About 35 discs were prepared for each sample and run for the test dose. The equivalent dose was calculated using Duller’s Analyst software.

Grun’s software was used for the age estimation.

Age of the sample = equivalent dose/annual dose rate

2.9 Mechanical and physical properties

The mechanical property was determined with the compression test performed on extracted samples, while the physical tests determining the water absorption, porosity and bulk density were carried out in accordance with IS 3495 (Part 1-4): 1992. The compression testing was carried out at a loading rate of 1.28 KN/sec [IS 3495 PART 1: 1992].

Water absorption (%) was calculated using formula $[M_2 - M_1] \times 100/M_1$.

Porosity (%) = $[(M_2 - M_1)/(M_2 - M_3)] \times 100$

Bulk density = $[M_1/(M_3 - M_2)] \times 10^3$,

where

M_1 = mass of an oven-dried specimen;

M_2 = mass of a saturated specimen;

M_3 = hydrostatic mass.

Compressive strength = maximum load at failure/net area of faces under compression.

2.10 Structural drawing

3D modelling was done to get the accurate dimensions and a perfect model of the well. It was done with the help of the H-BIM tool. The images required for the modelling were captured with a Kinect sensor and the Context Capture software was used for the modelling.

3 RESULTS AND DISCUSSION

3.1 Optical microscopy

The thin sections were examined with a polarising microscope and the pictures included in Figure 3 were

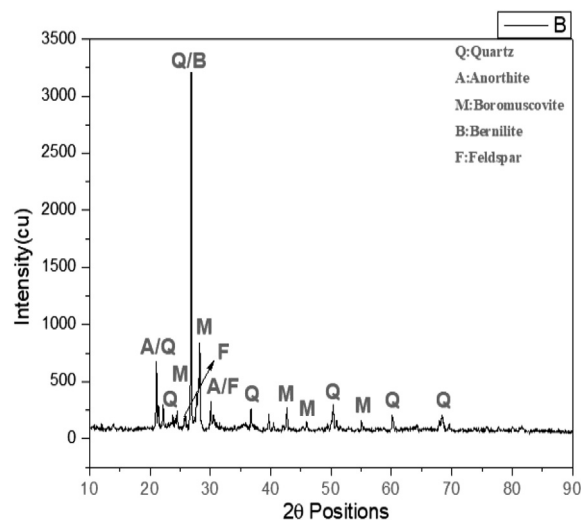


Figure 4: XRD of the brick sample

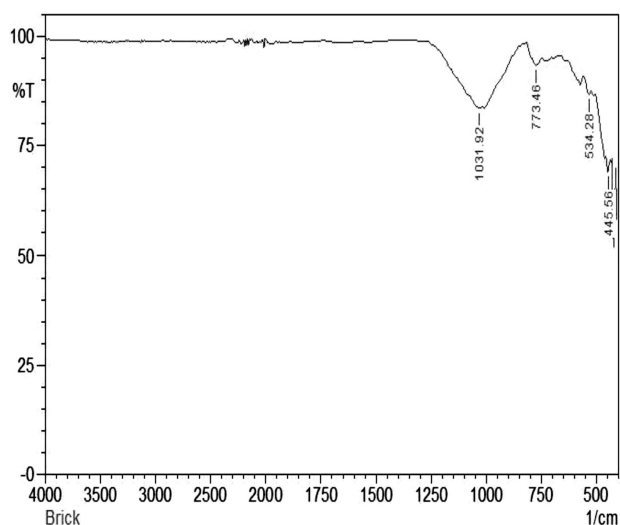


Figure 5: FT-IR peaks of the brick

obtained. In the polarized-light microscope, the bending of planar morphological elements such as rational crystal or cleavage indicate plastic crystal deformation.²⁵ From these images, the presence of quartz, feldspar, hematite and boromuscovite can be identified.²⁵

3.2 XRD

The XRD plots were recorded in a 2θ range of $10\text{--}90^\circ$. The XRD analysis of the brick sample in **Figure 4** indicates the presence of quartz, boromuscovite, berlinite, anorthite and feldspar. It also indicates the phase transformation of calcite into anorthite at different sintering temperatures.¹⁰ The XRD results are consistent with the FT-IR analysis which shows broadening peaks at $1100\text{--}1000\text{cm}^{-1}$, referring to the amorphization reaction.

3.3 FT-IR

Figure 5 depicts the FT-IR spectra of the brick sample. The deconvolution of the peak range between $2450\text{--}2000\text{cm}^{-1}$ and $1200\text{--}800\text{cm}^{-1}$ indicates the exposure of the brick sample to different firing temperatures.⁹ The broad absorption band at 1031.92cm^{-1} is of an asymmetric Si-O stretching followed by a strong absorp-

tion band at 445.56cm^{-1} , also indicating the presence of quartz.^{9,10} The highly sensitive carbonate band, which normally shows at around 1400cm^{-1} for a sintered sample was not found for this sample, indicating the transformation of calcite to anorthite.¹⁰ The peak at 534.28cm^{-1} corresponds to the Fe-O bending of hematite.¹⁴

3.4 FE-SEM

The scanning-electron-microscope observations provided valuable information about the internal morphology, distribution matrix, pore-structure formation and size of the crystals.¹⁴ This sample contains pores, which are of a medium size, and it is less vitreous in nature, indicating that the temperature, at which the sample was fired was less than 950°C .²³ The presence of feldspar and the accumulation of quartz crystals that are in agreement with the XRD and FT-IR results are shown in **Figure 6**.

3.5 XRF

Table 1 lists the XRF-analysis results confirming that the brick sample contains high concentrations of SiO_2 , Al_2O_3 and very low amounts of Na_2O , MgO and CaO . Since the amount of calcium present is lower than 6 %, the clay used for the manufacturing of bricks was non-calcareous.^{6,14} Since the amounts of inorganic oxides such as Fe_2O_3 , MgO , CaO , etc. are higher than 9 % by weight, they indicate fluxing and a medium refractory property.¹⁰ The presence of Na, Ti, Ca, Mg and K has a non-negligible effect on the hardness of the brick.⁹

Table 1: XRF data

Oxides	Mass %
SiO_2	66.9
Al_2O_3	17.2
Fe_2O_3	4.3
CaO	3.58
K_2O	3.13
Na_2O	2.12
MgO	1.27
TiO_2	0.748
P_2O_5	0.297

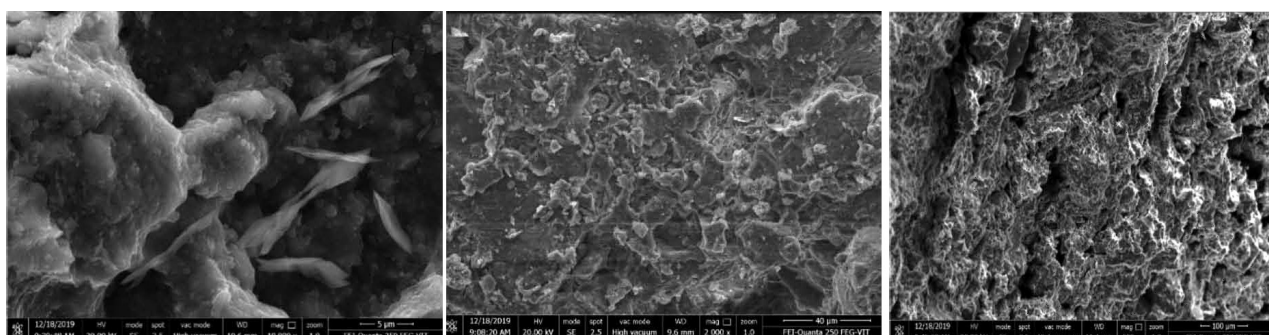


Figure 6: FE-SEM images: a) quartz accumulation; b) feldspar; c) porous structure

3.6 TGA

The TGA curve shown in **Figure 7** indicates a negligible weight loss in the region of 600–900 °C, which indicates the dihydroxylation of the raw materials for the brick at the firing temperature, containing a low amount of calcite according to the XRD analysis.⁶ The weight loss of around 1.5 % occurs in the temperature range of 200–400 °C due to the loss of interlayer water and clay dihydroxylation.^{2,13} The total weight loss is less than 2 % by weight. The firing temperature of the sample is mostly between 750–900 °C. The absence of mullite and cristobalite indicates that the firing temperature is below 950 °C.^{6,24} The presence of anorthite indicates the decomposition of calcite at higher firing temperatures.² Between 700 and 800 °C, the clay is of an orange-red color.¹⁴ DTA (**Figure 8**) shows the physical changes that do not involve weight changes such as crystallisation, melting, the change in the solid phase and homogeneous reaction in the solid state.

3.7 OSL

OSL is a widely used technique in archaeological dating.²¹ The single-aliquot-regeneration (SAR) protocol of Wintle and Murray is used for the determination of the equivalent dose/palaeodose. The amount of the light emitted during the luminescence measurement of the sample depends upon the total radiation dose, to which the crystalline material was exposed while it was buried and it is called the natural signal. This measured signal provides the measure of the palaeodose. The total radiation dose, which a material receives annually is called the annual radiation dose. Based on the OSL dating, the approximate age of the sample is determined to be 2100 years. This test confirms that the well was constructed in the Early historic period.

3.8 3D model

The modelling of the brick well was based on the principle of photogrammetry, which is a part of H-BIM.²² This is a technology used for obtaining infor-

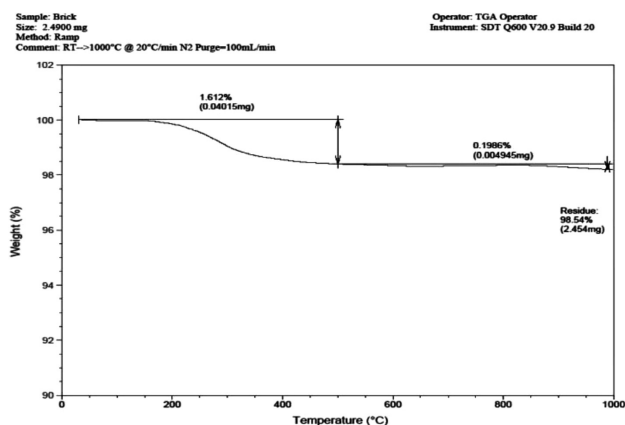


Figure 7: TGA curve of the brick sample

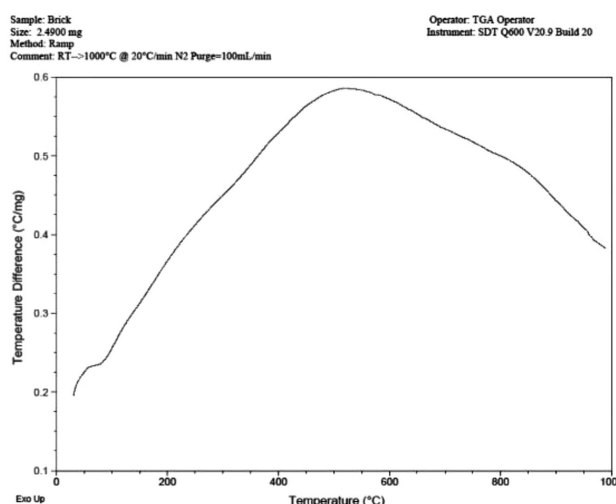


Figure 8: DTA curve of the brick sample

mation about objects through a process of measuring recorded images. The final output is obtained after going through various stages such as the collection of images, archaeological relevance or survey data. The images were captured at an angle of 30° and stitched together with an overlap of 30°. The rendered 3D model of the brick well is shown in **Figure 9**.

3.9 Physical and mechanical properties

The water absorption and porosity are the key factors in determining the durability of a brick sample. The water absorption is relatively proportional to the porosity. The value obtained for our sample is 15 %; therefore, the sample is durable. The standard water-absorption value must be less than 21 % for high-quality bricks.⁹ The value of porosity obtained is 22.8 % and the bulk density is 1.873 kg/m³. The compressive strength of the brick obtained is 6.7 MPa. A compressive strength between 2.5 MPa and 7 MPa is recommended for horizontally perforated bricks used for indoor constructions.¹⁰ The values are compared with respect to the IS Codes.

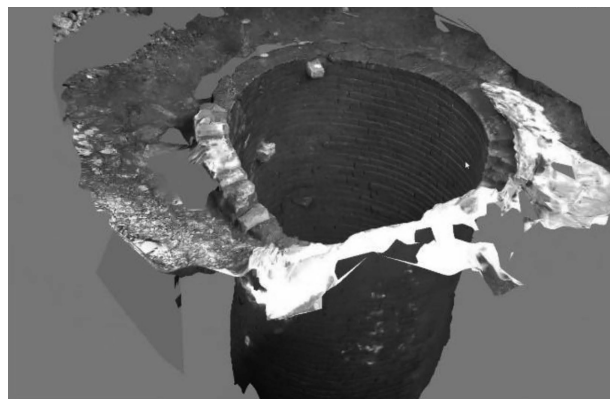


Figure 9: 3D rendering of the Pattaraiperumbudur brick well

4 CONCLUSION

From the tests conducted, it is concluded that the bricks used in the construction of the well exhibit an optimum water absorption, compressive strength and slightly porous structure. The combined results of XRD, OM, SEM, TGA confirm the presence of minerals such as quartz, feldspar, anorthite and boromuscovite. They also provide an idea of the production technology used, indicating that the bricks are non-vitreous in nature and produced from Ca-poor clay with high amounts of silica and alumina at a firing temperature of less than 950 °C. The clay used in the production of exhibits inorganic fluxing and a medium refractory property. The suitable proportion of quartz in the clay prevents shrinkage and warping and improves the durability of bricks.⁹ The OSL dating technique gives the age of the sample, confirming that the brick well was constructed in the Iron Age period.

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